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periodic table to

determine how

many protons,

neutrons, and

electrons are

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present in each of
the following atoms.
Write your answers
in the spaces

provided. Atom 12.

iodine-125 13.

potassium-39 14.

iron-56 Protons

Neutrons Electrons

Write the chemical
symbol for each of
the ions described
below.

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to one of the gas

laws 1. The

pressure Of a gas is

inversely

proportional to its

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volume. 2. The sum
of the partial
pressures of gases
in a mixture is
equal to the total
pressure of the
mixture.

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of objects that
move through the
air, is affected by
speed, size, and
shape 8. net force
9. microwelds 10.
rolling 11. air

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12. acceleration 13.

sliding 14.

parachute 1.

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that each of the

following atoms will

acquire when it has

a complete set of

valence electrons.

23. O 24. Na 1 25.

F 27. Ca 28. 29.

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radius. – -mc
nuc,lcus 30. Why do
atoms get smaller
as you move across
a period? and Cloud
31.

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echoing effect

caused by many
reflections of

sound. 3 1 Review

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iodine-126

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remaining in a
sample originally
containing 26.3 g of
iodine-126. The

half-life of

iodine-126 is 13

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