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~~Alcohols And Ethers~~

11.27. Preparation of ethers
by acid dehydration of
secondary or tertiary
alcohols is not a suitable

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method. Give reason. Ans:
Acid catalysed dehydration
of primary alcohols to
ethers occurs by SN 2
reaction involving
nucleophilic attack by the
alcohol molecule on the
protonated alcohol molecule.

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atom in hydrocarbon is
replaced by -OH group.

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Chapter 11 - Alcohols,
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325 Solution 1 Primary
alcohol ' (i), (ii), (iii)
Secondary alcohol ' (iv),
(v)

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The formation of ethers by dehydration of alcohol is a bimolecular reaction (S N 2) involving the attack of an alcohol molecule on a protonated alcohol molecule. In the method, the alkyl group should be unhindered.

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In case of secondary or tertiary alcohols, the alkyl group is hindered. As a result, elimination dominates substitution.

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Chemistry Chapter-11:
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names of the compounds given

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below : Solution : (i) 2, 2, 4 -Trimethylpentan - 3 - ol
(ii) 5 - Ethylheptane - 2, 4 - diol (iii) Butane - 2, 3 - diol (iv) Propane - 1, 2, 3 - triol (v) 2-Methylphenol (vi) 4 - Methyl phenol (vii) 2, 5 - Dimethylphenol

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Ethers NCERT Intext

Exercises. Question 1.

Classify the following as primary, secondary and tertiary alcohols : Answer:

(i) 1° , (ii) 1° , (iii) 1° ,
(iv) 2° , (v) 2° , (vi) 3° .

Question 2. Identify allylic

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alcohols in the above
examples. Answer:

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In cleavage of ethers,
higher alcohol and lower

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alkyl halide is formed
except in tert. butyl methyl
ether, in which $(\text{CH}_3)_3\text{C}^+$
is more stable so $(\text{CH}_3)_3\text{C}-$
 I and CH_3OH is formed.
Anisole with HI gives phenol
and CH_3I because phenoxide
ion is stabilised by

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Phenols Alcohols and phenols

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are formed when a hydrogen atom in hydrocarbon, aliphatic and aromatic respectively, is replaced by hydroxyl group (-OR group).
Classification of Alcohols and Phenols In alcohols, -OR group is attached to Sp^3

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phenols may be classified as monohydric, dihydric, trihydric or polyhydric according to number of hydroxyl groups they contain one, two, three or many respectively in their molecules. 2.

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Alcohols and phenols are
formed when a hydrogen atom

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in a hydrocarbon, aliphatic and aromatic respectively, is replaced by -OH group. The substitution of a hydrogen atom in a hydrocarbon by an alkoxy or aryloxy group (R-O/Ar-O) yields another class of

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compounds known as 'ethers',
for example, CH_3OCH_3
(dimethyl ether).

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In case of secondary or tertiary alcohols, the alkyl group is hindered. As a result, elimination dominates over substitution.

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Alcohols, Phenols and Ethers

Alcohols, Phenols and Ethers

Important Questions.

Alcohols, Phenols and Ethers

Very Short Answer Type

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Questions. Question 1. What is Ether? Answer: Compounds formed by the substitution of hydrogen atom of hydrocarbon by alkoxy group are called Ethers. Question 2.

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