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CHAPTER 17 REVIEW Reaction  
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ANSWER Answer the following  
questions in the space provided.

1. The reaction for the  
decomposition of hydrogen  
peroxide is  $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ . List three ways to

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speed up the rate of decomposition. For each one, briefly explain why it is effective, based on collision theory.

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CHAPTER 12 & 13: SOLUTIONS

The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions,

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