

Principles Of Electronic Materials Devices 3rd Edition Solutions

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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q1 and Q2 separated by a distance r is given by QQ PE = 4πε o r First we consider the interaction between Na+ ion and 6Cl- ions at distance r Applying Coulomb's law we have Q1Q2 (-6e)(+e) - 6e PE1 = = 4πε o r 4πε o r 4πε o r Similarly, we now consider 12 Na+ ions as second nearest neighbors at a distance r QQ (+12e)(+ e) 12e = PE2 = = 4πε o r 4πε o r 4πε o r and Cl ...

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