

Solutions Manual Chemical Equilibrium Chemistry

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$$\text{H}_2(g) + \text{I}_2(g) \rightleftharpoons 2\text{HI}(g)$$

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Solutions Manual Chemical Equilibrium Chemistry Chemical equilibrium is the state of a system in which the rate of the forward reaction is equal to the rate of the reverse reaction. [Figure](#)
$$\text{H}_2(g) + \text{I}_2(g) \rightleftharpoons 2\text{HI}(g)$$
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Nelson Chemistry 12 Solutions Manual Unit 2 Chemistry 12 Unit 2 Notes - Equilibrium Once this has happened for awhile, there is a build up of NO2 molecules in the same flask Once in awhile, two NO2 molecules will collide with each other and join to form a

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A solution equilibrium occurs when a solid substance is in a saturated solution. At this point, the rate of dissolution is equal to the rate of recrystallization. Although these are all different types of transformations, most of the rules regarding equilibrium apply to any situation in which a process occurs reversibly.

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